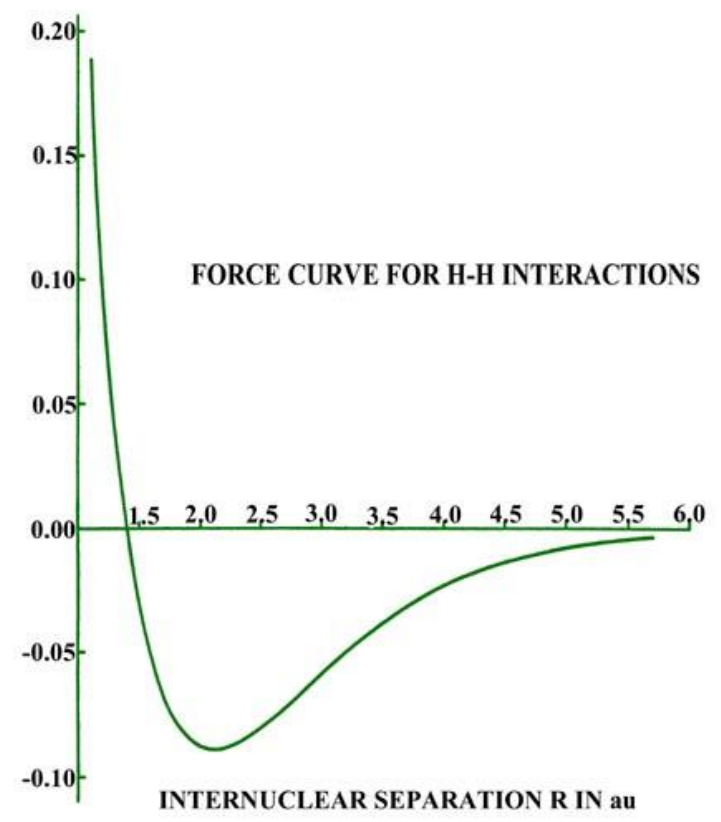
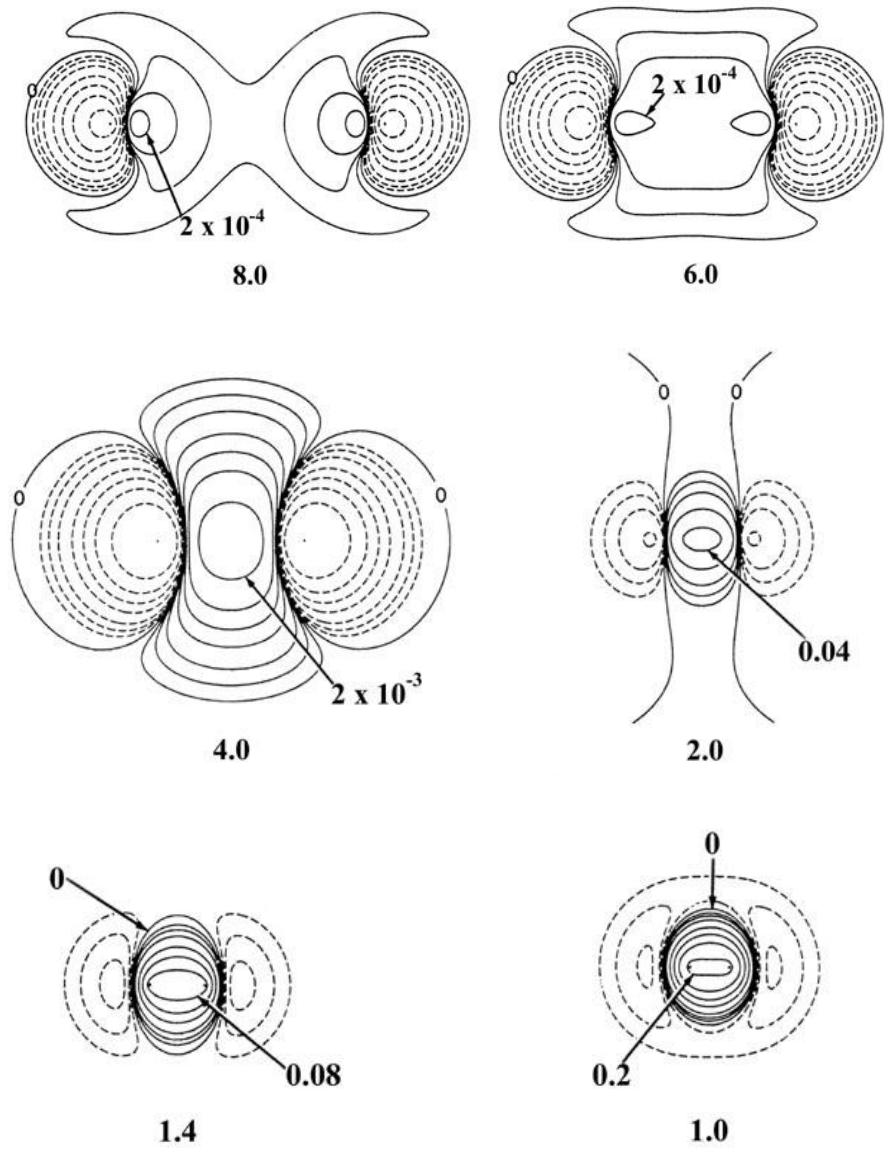


Scope of the Lecture

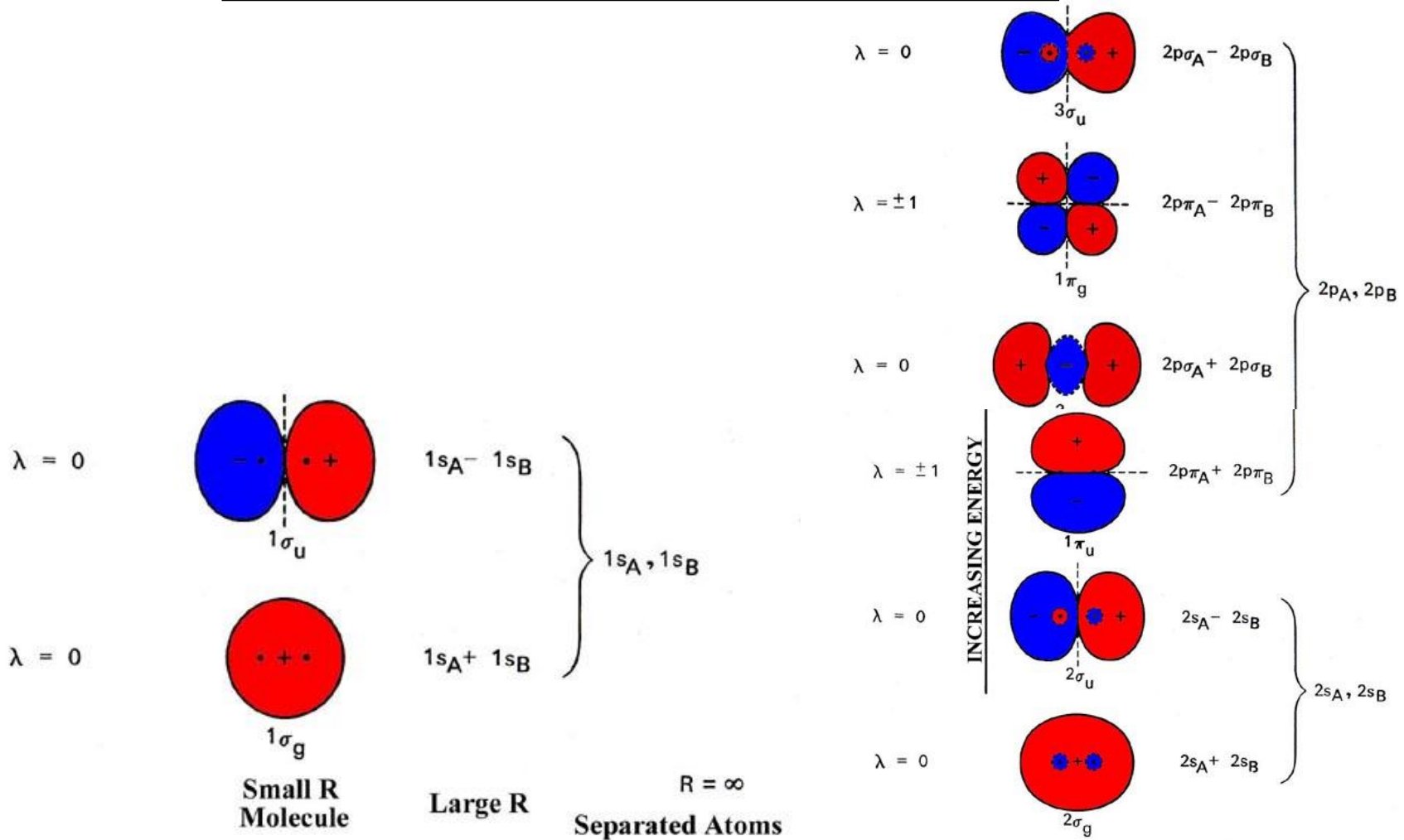
1. Molecular gain media
2. Vibronic, vibrational and rotational transitions
3. Energy bands in bulk semiconductor
4. Bulk density of states
5. Electron concentration
6. Selection rules for electronic transition in semiconductor
7. Absorption and emission
8. Necessary condition for gain
9. Recombination channels
10. Electronic states in quantum wells
11. Quantum wires, quantum dots

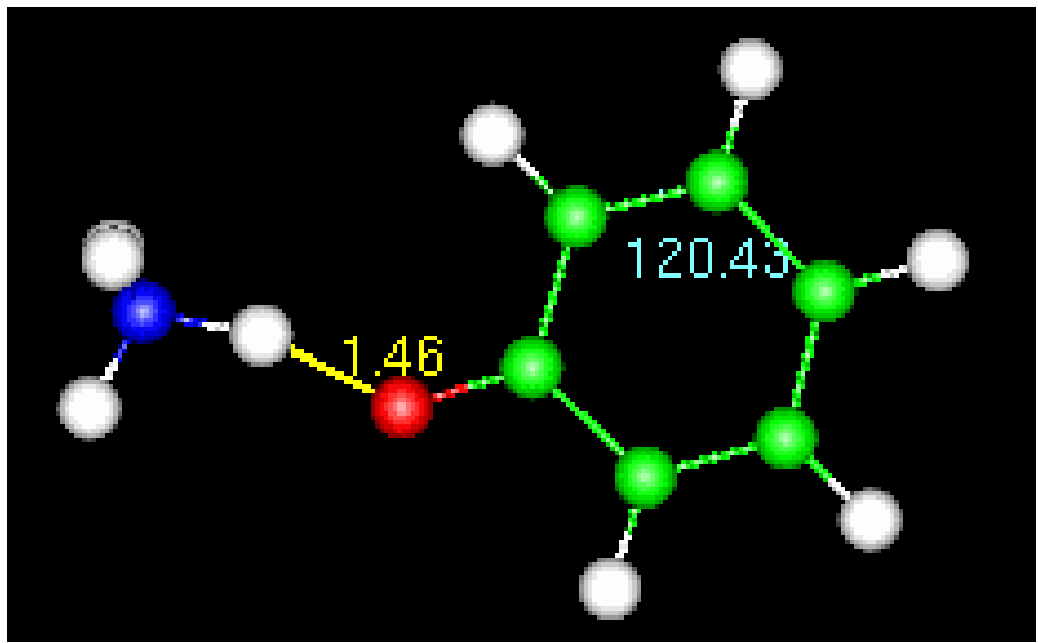
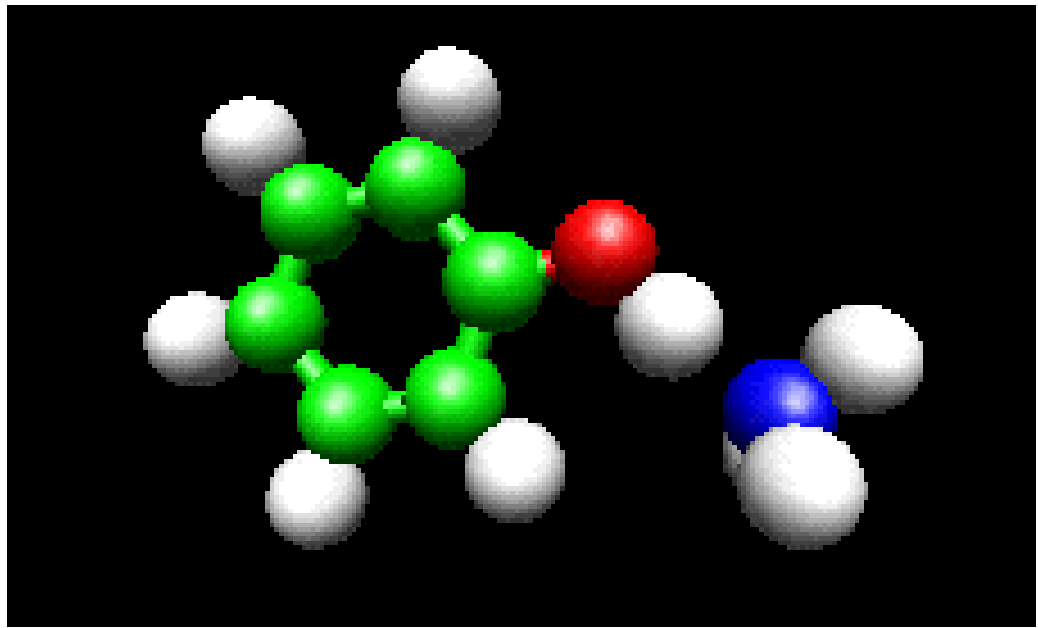
Reading: Ch. 3

Formation of H₂



Orbitals in homonuclear diatomic molecules





Electronic levels in N₂ molecule

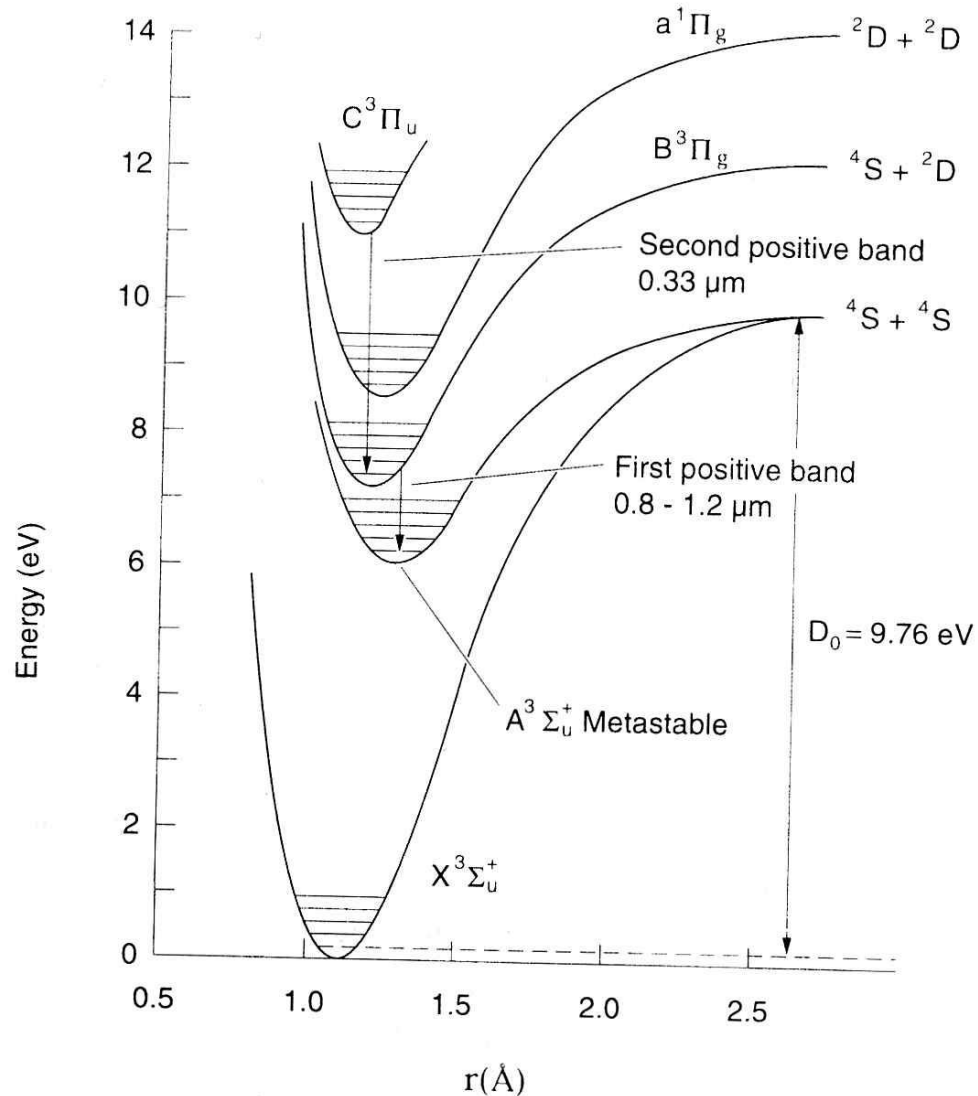


Figure 5-7 Energy levels of molecular nitrogen as a function of separation distance r between the two nitrogen atoms and also the dissociation energy D_0

Vibronic, vibrational-rotational, rotational transitions

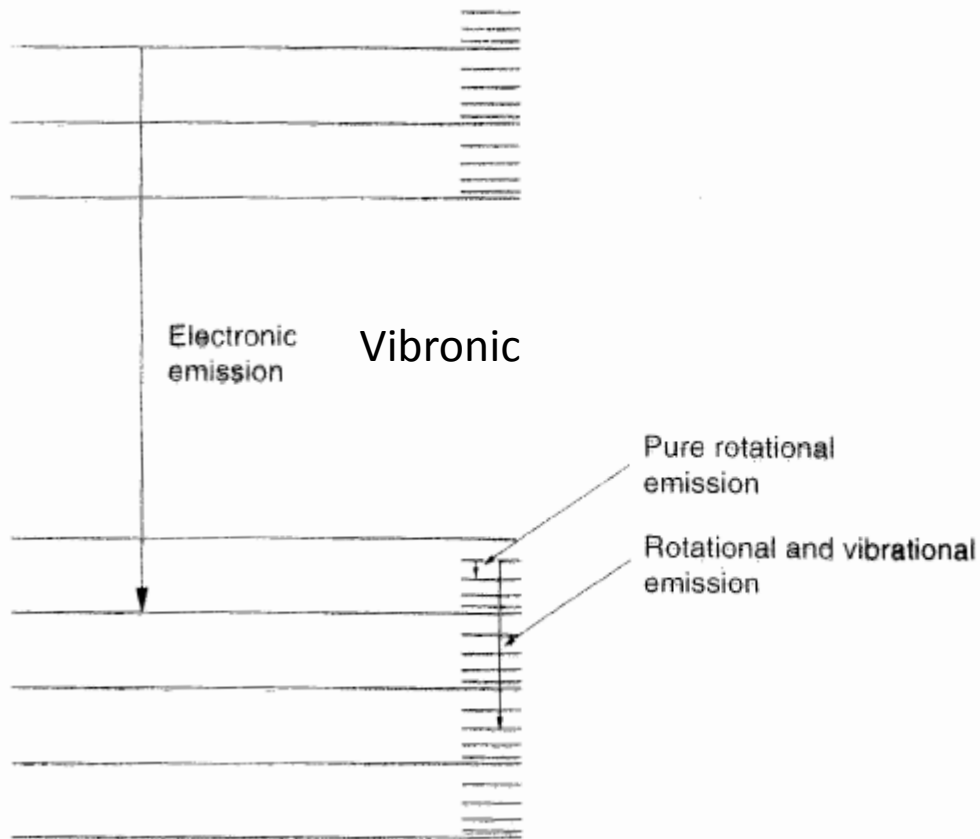


Figure 5-2 Energy levels of a molecule, indicating electronic emission as well as pure rotational emission and rotational-vibrational emission

Franck-Condon principle

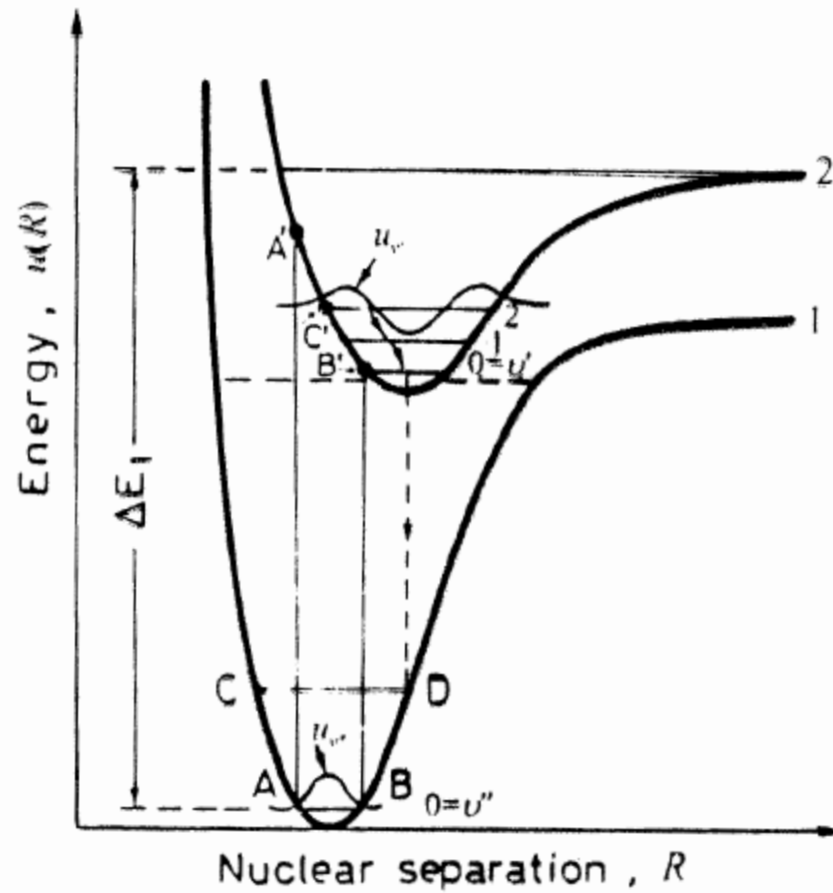


FIG. 3.6. Allowed vibronic transitions for a diatomic molecule.

Stokes law

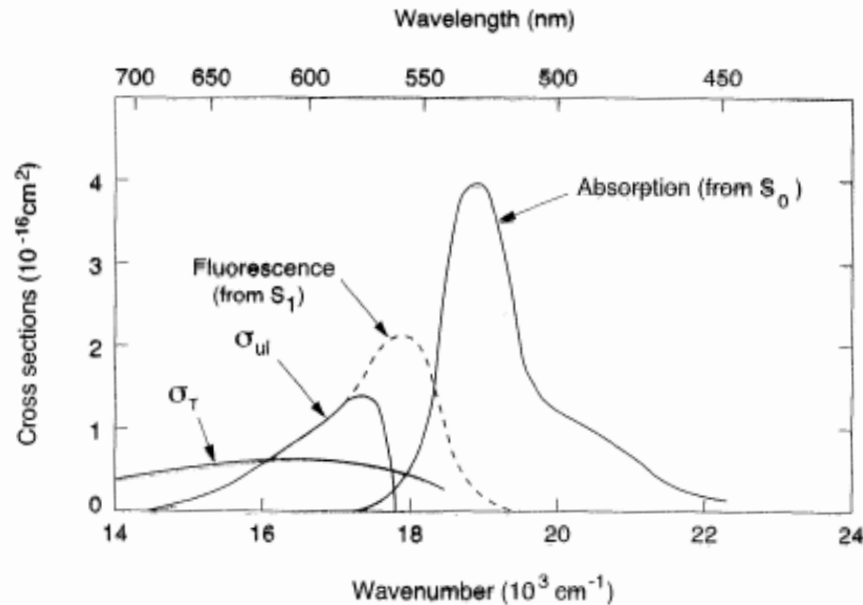
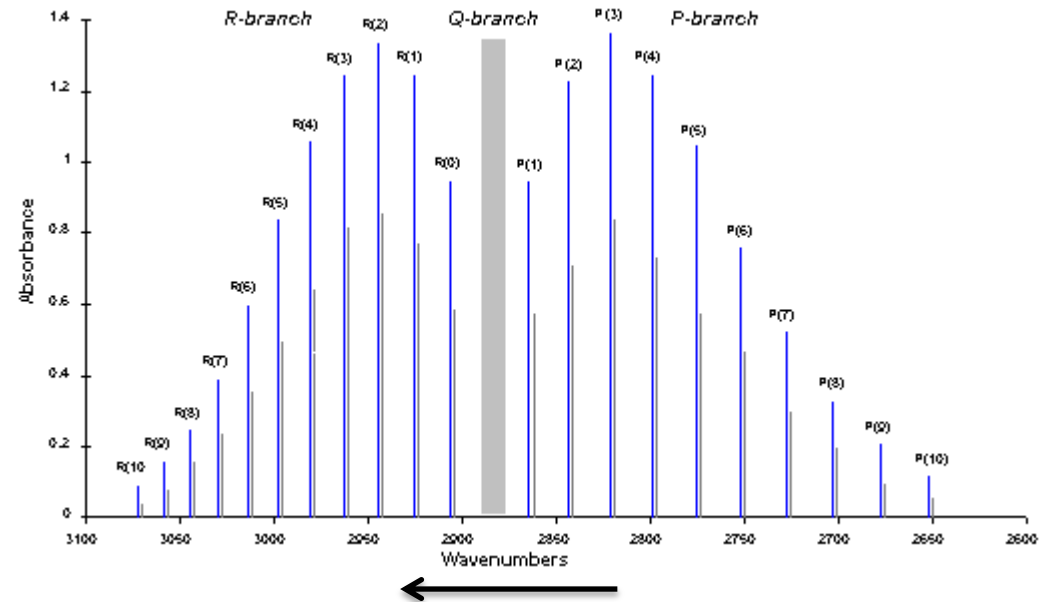
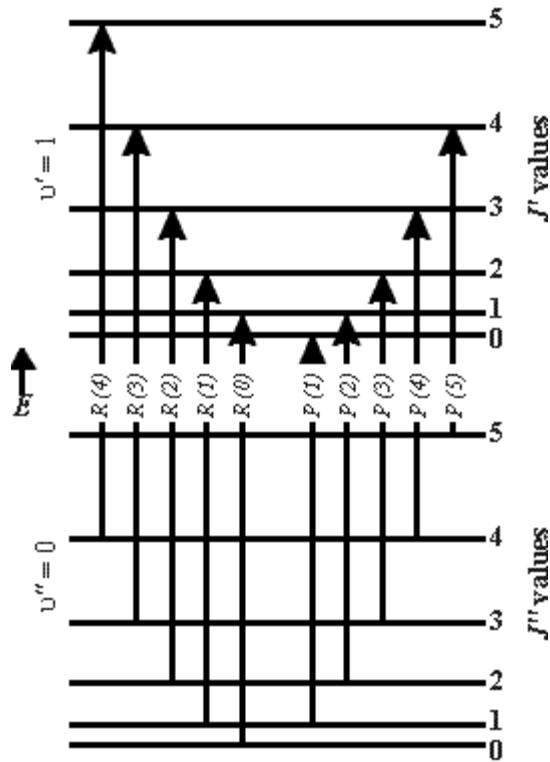


Figure 9-17 Absorption processes associated with absorption from S_0 and absorption from T_1 (σ_T), where wavenumber is the reciprocal of wavelength

GROUND-STATE ABSORPTION IN DYE LASERS Organic dye lasers have an overlap between the absorption (pumping) spectrum and the emission spectrum, as shown in Figure 5-10 for the dye RhB. This overlap is determined by the relationship of the singlet ground-state S_0 and excited-state S_1 energy levels shown in Figure 5-12.

Ro-vibrational transitions



Selection rules :

$$\Delta v = \pm 1$$

$$\Delta J = 0, \pm 1$$

Spectral branches :

$\Delta J = 0$ Q-branch (might be absent)

$\Delta J = +1$ R-branch

$\Delta J = -1$ P-branch

Population of rotational states

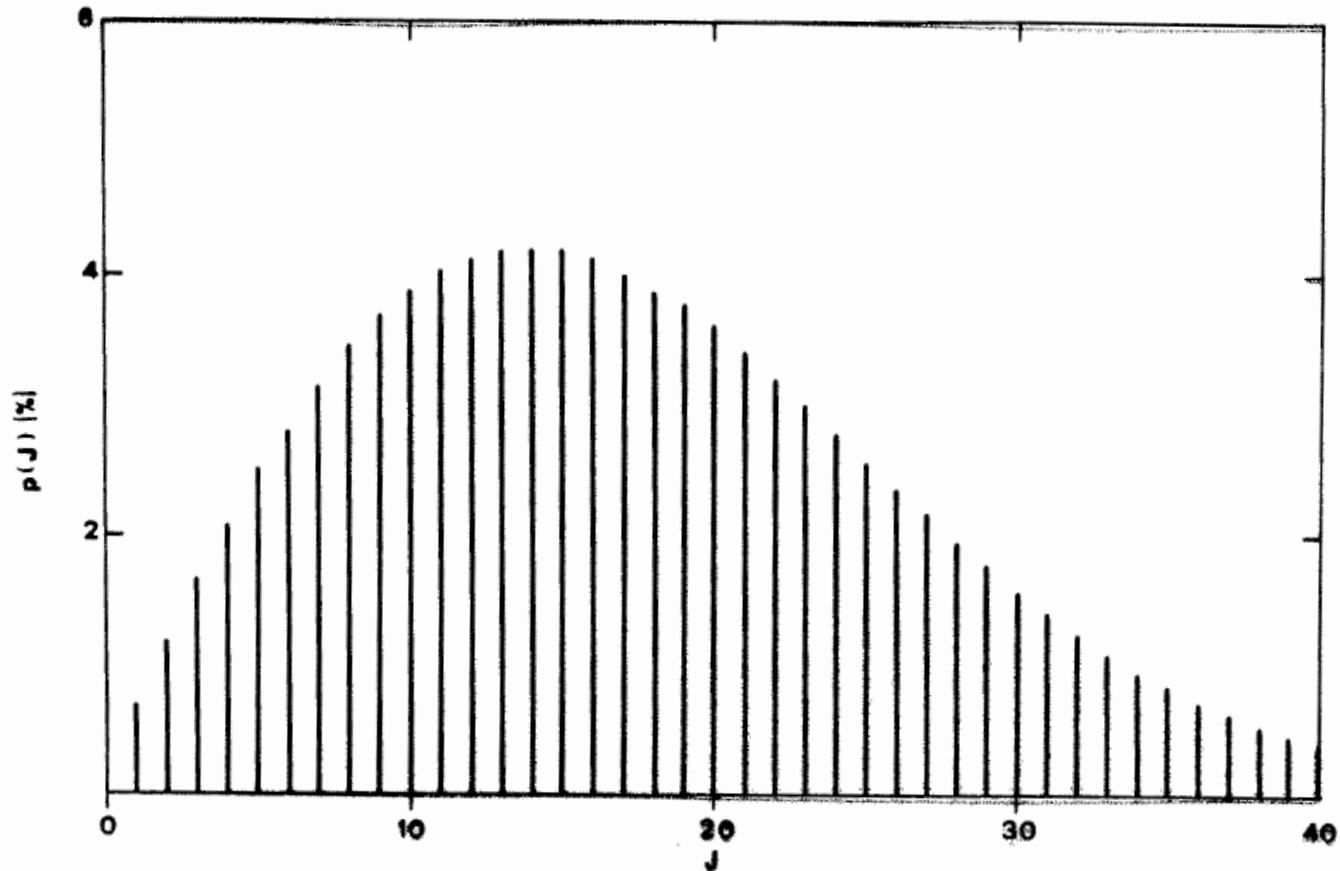


FIG. 3.5. Population distribution among rotational levels of a given vibrational state.

Formation of solids

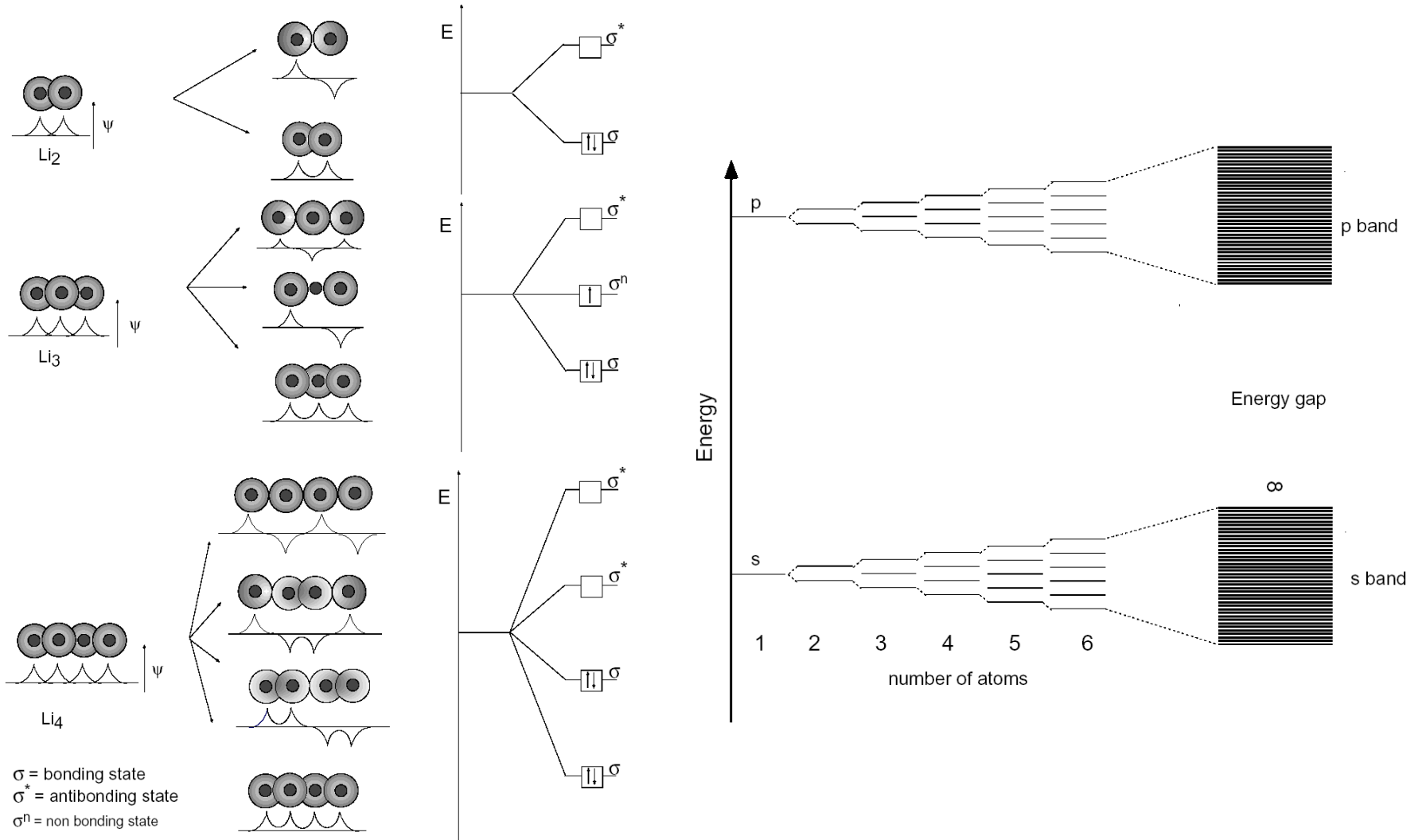
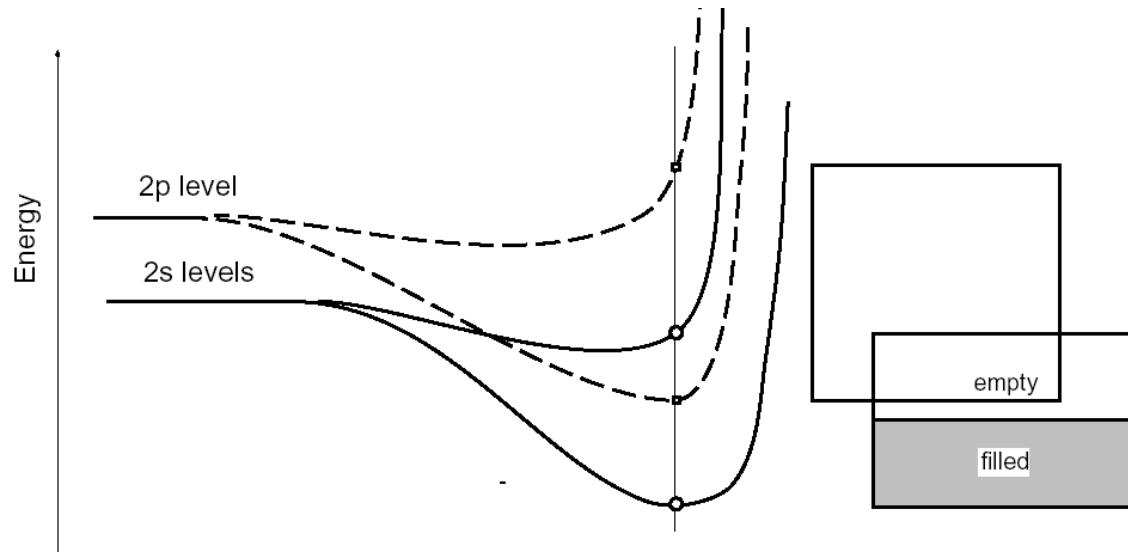


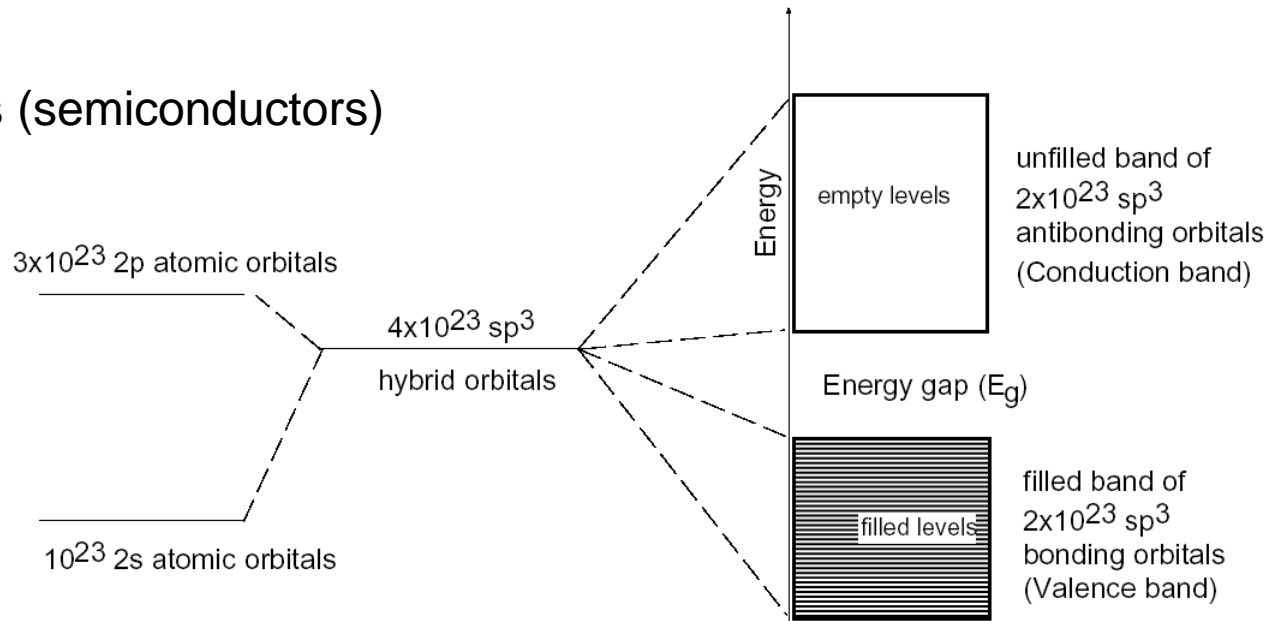
Figure 3 Formation of molecular chains of lithium atoms.

Formation of energy bands in solids

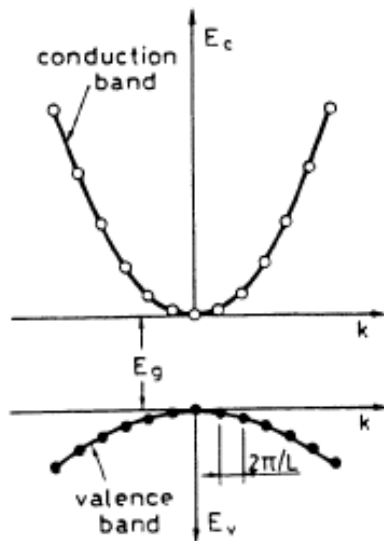
Metals



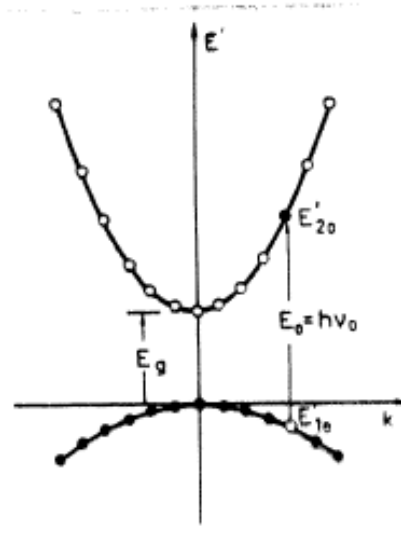
Dielectrics (semiconductors)



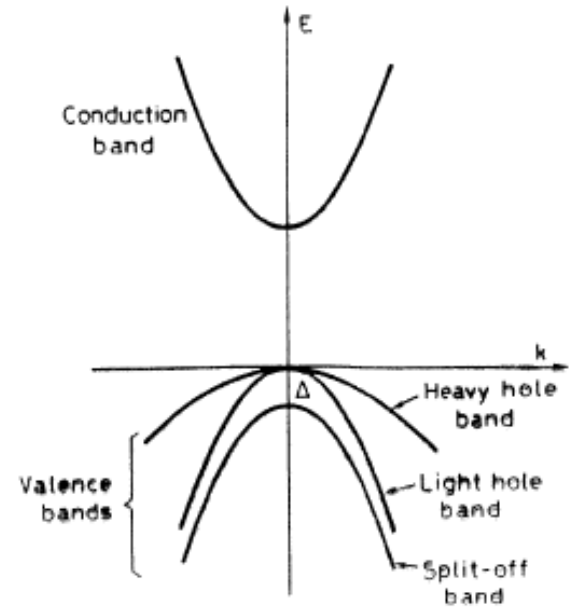
Electronic bands in direct-bandgap semiconductor



(a)

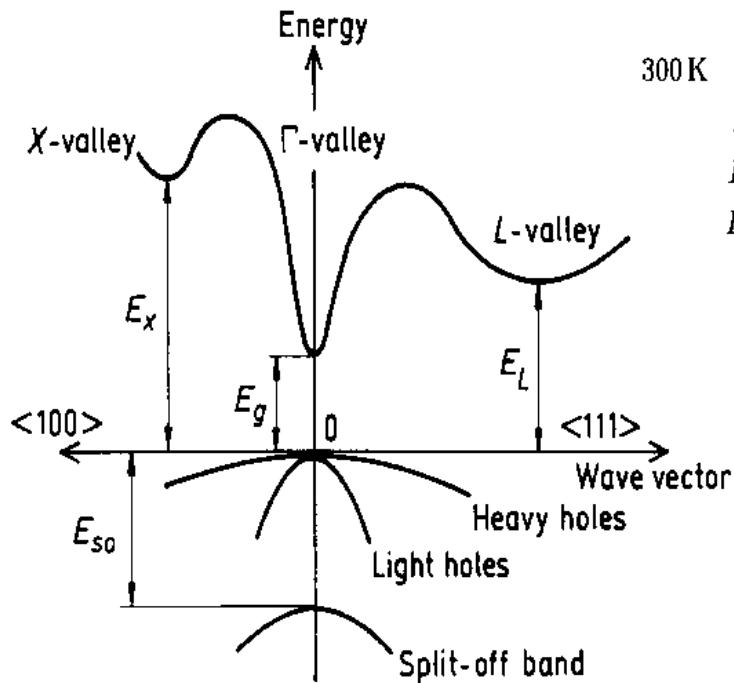


(b)



Direct and indirect bandgap semiconductors

GaAs (InAs, InP, InSb, etc.)



300K $E_g = 1.42$ eV
 $E_L = 1.71$ eV
 $E_x = 1.90$ eV
 $E_{so} = 0.34$ eV

Si (Ge, GaP, etc.)

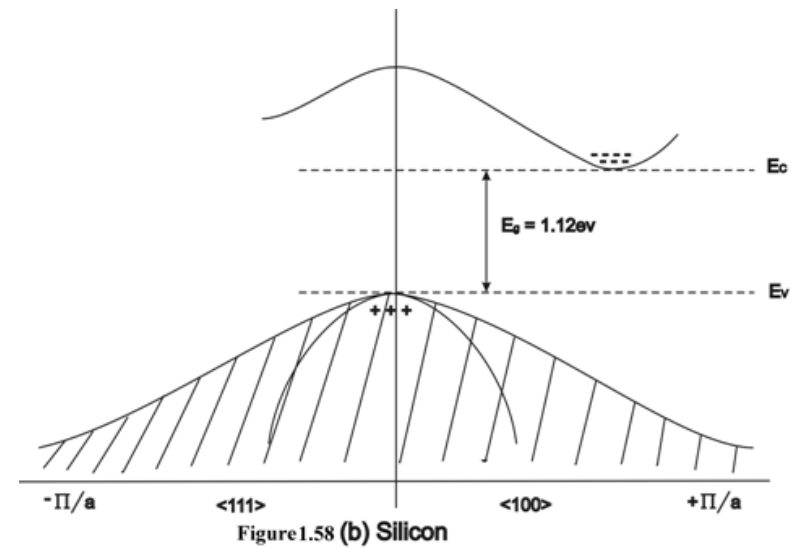


Figure 1.58 (b) Silicon

Fermi-Dirac distribution

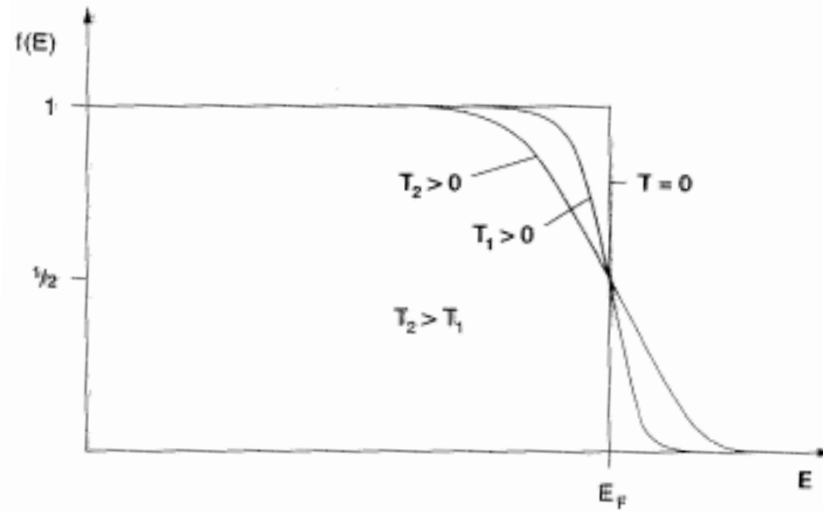


Figure 5-21 Fermi-Dirac distribution function

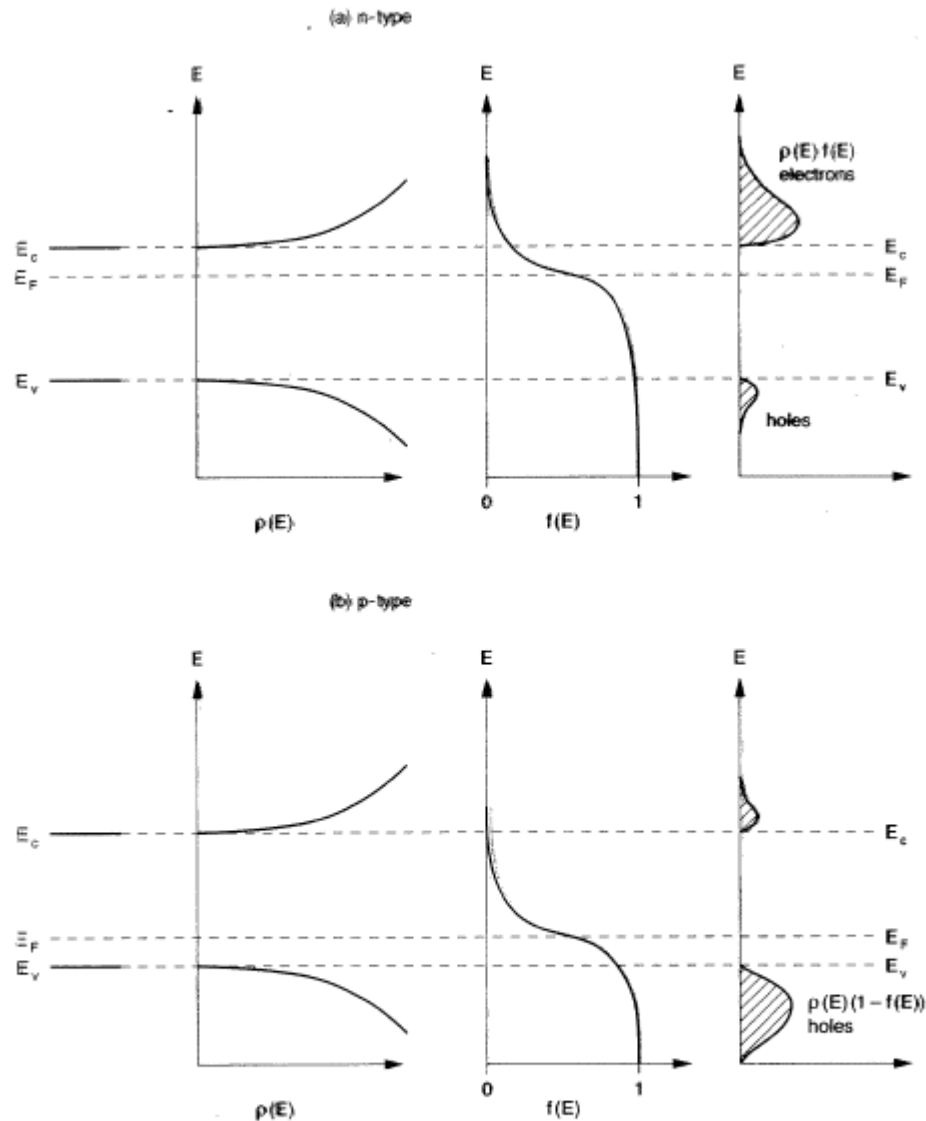
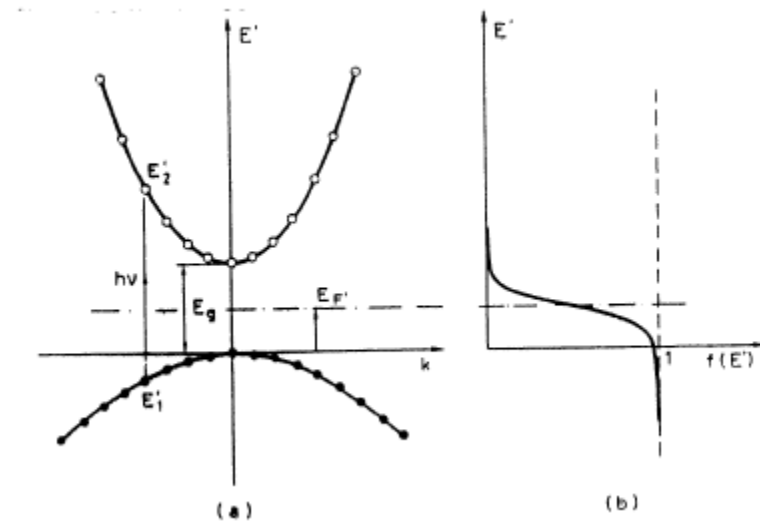
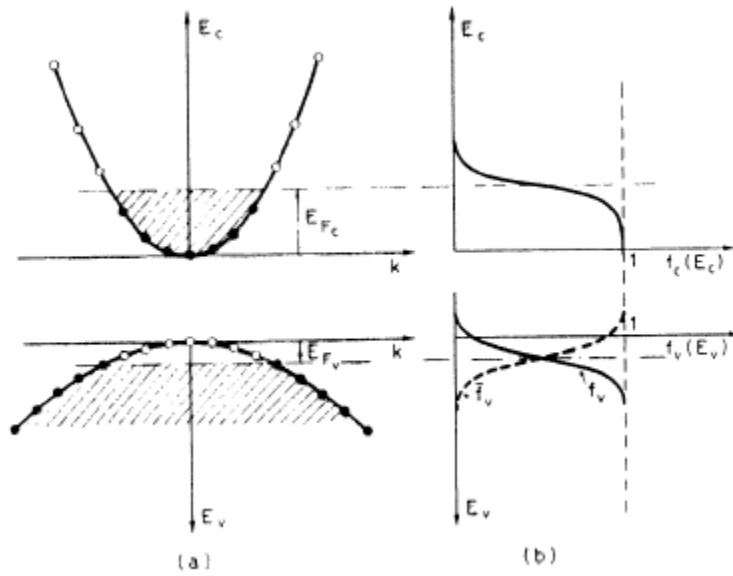


Figure 5-24 Carrier distribution in the conduction band and valence band for an extrinsic semiconductor for (a) n-type doping (excess electrons) and (b) p-type doping (excess holes)

Fermi level and quasi-Fermi levels



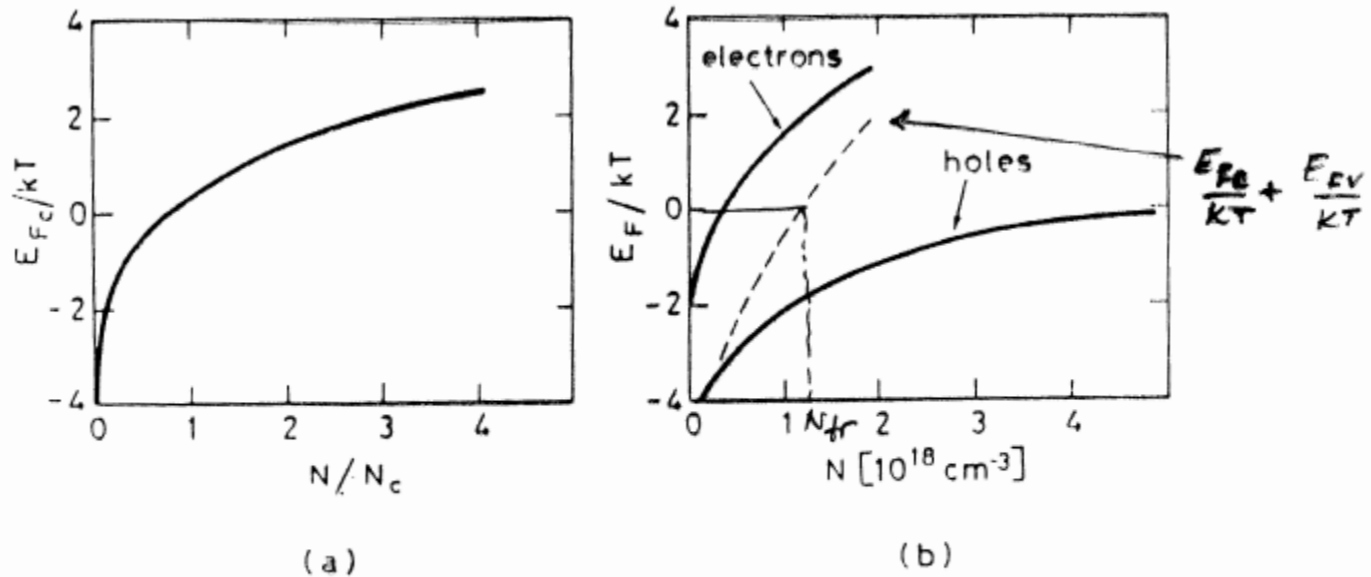
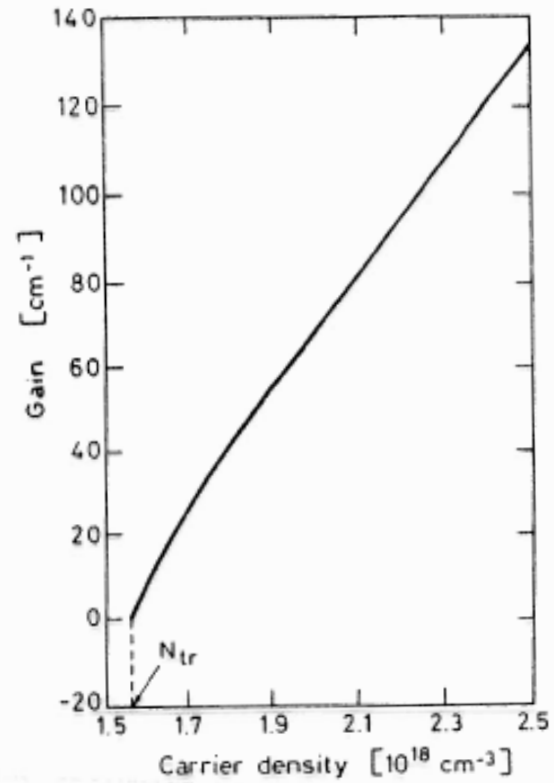
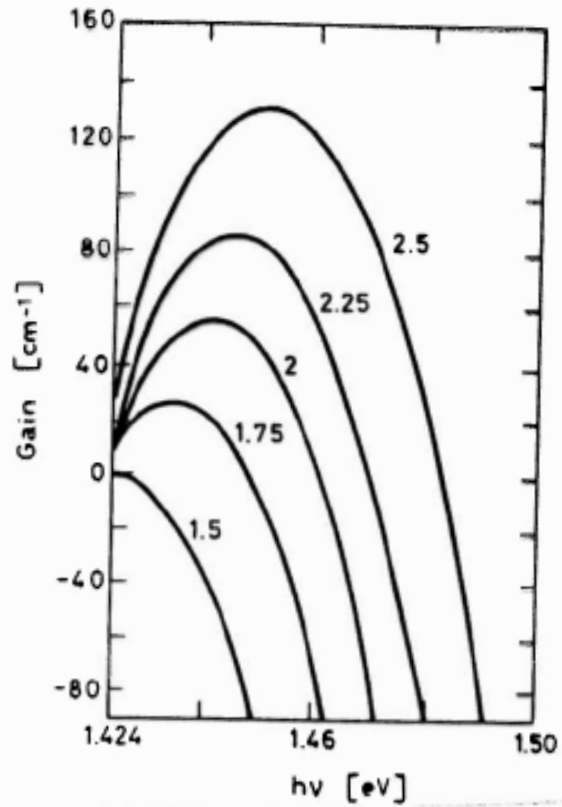


FIG. 3.15. (a) Normalized plot of the quasi-Fermi energy of the conduction band E_{F_c} versus the normalized concentration of injected electrons, N . The same normalized relation also holds for holes in the valence band. (b) Normalized plots of the quasi-Fermi levels of both valence and conduction bands E_F/kT versus the concentration of injected carriers N for GaAs.

Optical gain in semiconductors



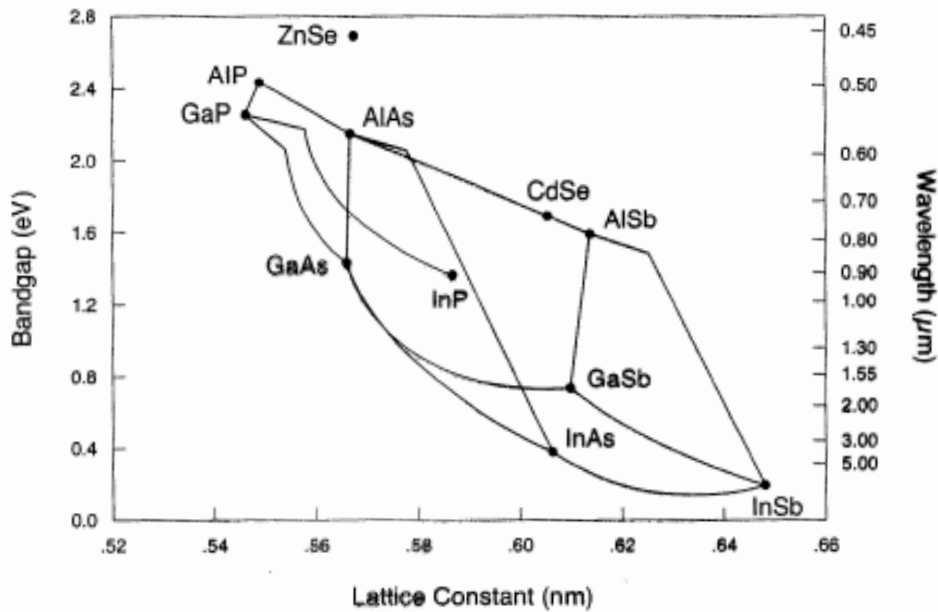
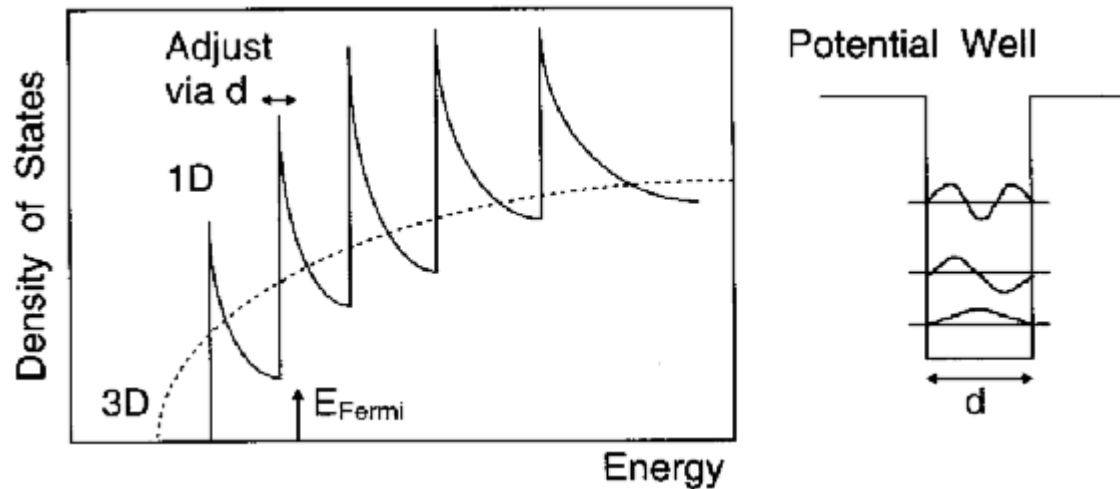
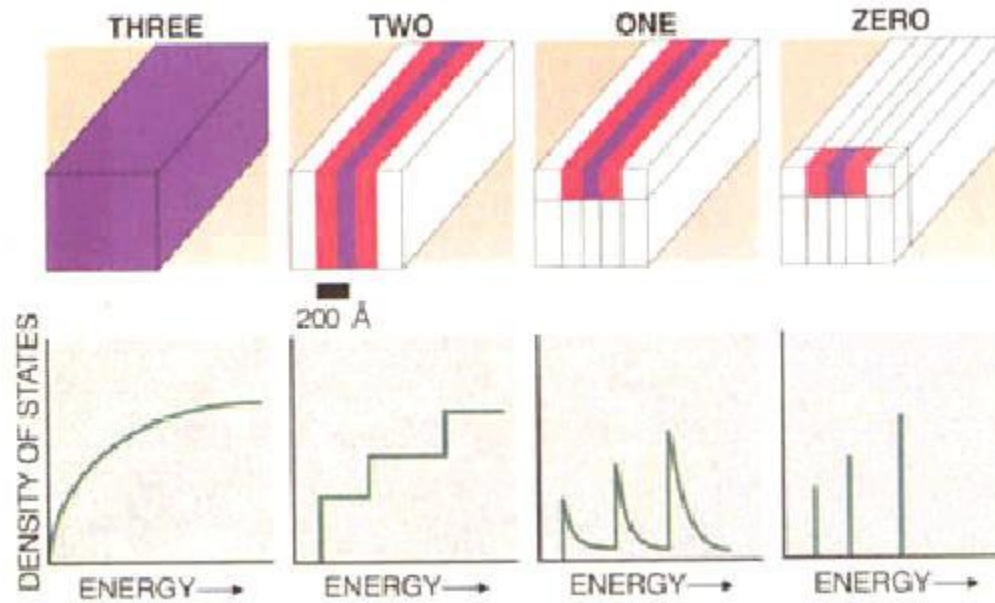


Figure 5-31(a) Potential laser wavelengths for semiconductor laser materials with bandgaps ranging from 0.2 eV to 2.8 eV

Mesoscopic structures

Quantum wells - 2D, Quantum wires – 1D, Quantum dots – 0D



Main Keywords

Molecular orbitals

Vibrational and rotational degrees of freedom

Frank-Condon principle

Stoke's law

Conduction and valence bands

Electron and hole effective mass

Density of states

Fermi level, Quasi-Fermi levels

Joint density of states

Transparency carrier density

Recombination: radiative, nonradiative, Auger

Subbands in quantum wells

Strained quantum wells

Quantum wires, quantum dots

Problems

3.7, 3.8, 3.9, 3.10

Examples: 3.5 (error correction: $m_v = m_{hh}$), 3.6, 3.7, 3.10