

Scope of the Lecture

1. Stimulated transition rates, x-sections, line-shapes
2. Nonradiative decay
3. Resonant energy transfer
4. Strongly coupled levels
5. Gain and absorption saturation
6. Spectral and spatial hole-burning
7. Amplified spontaneous emission

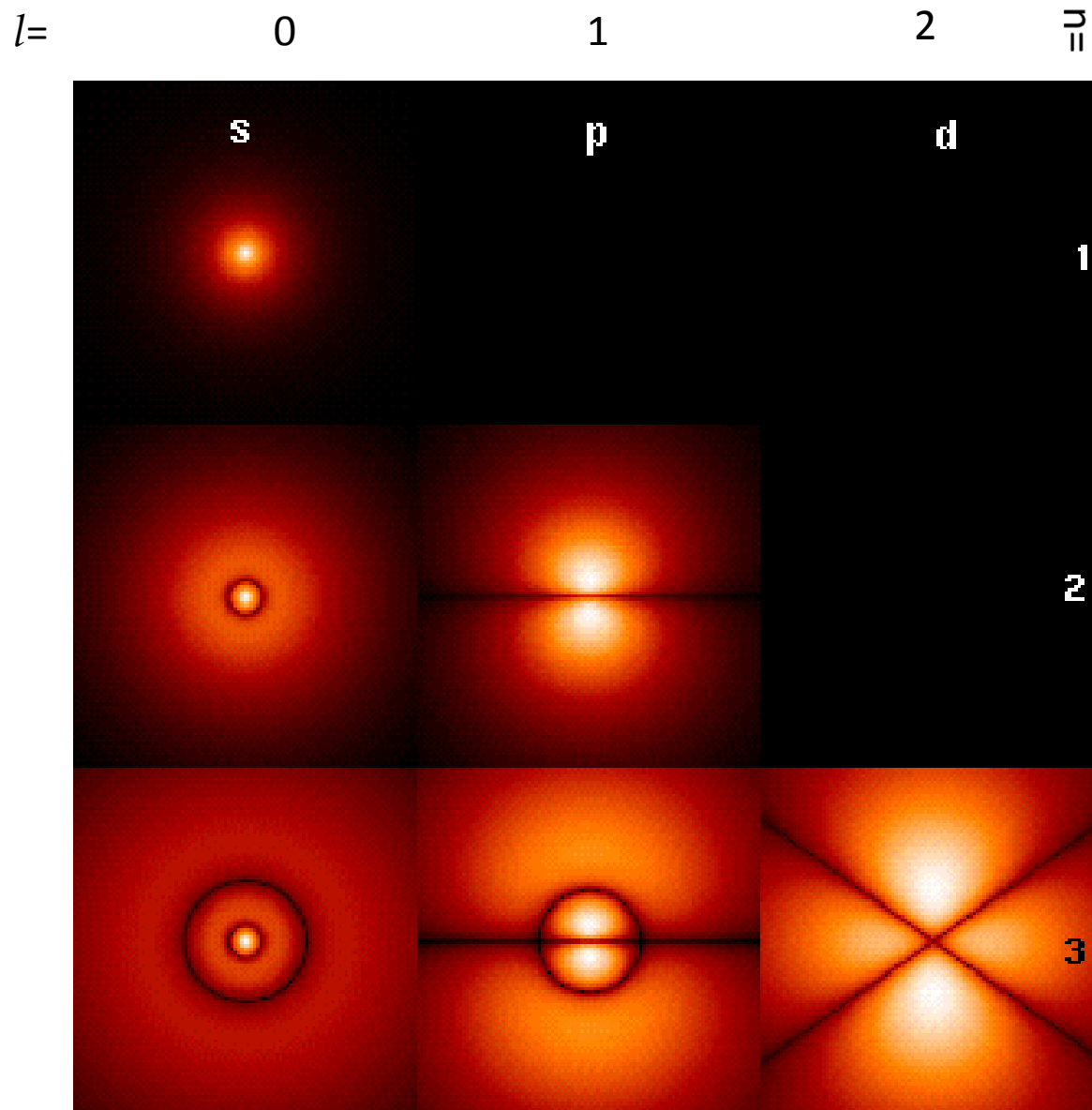
Reading: Ch 2.6, 2.7, 2.8, 2.9*

Electronic configurations

| | | | | | | | | | | | | | | | | | |
|------------------------|--------------------|--------------------------|--------------------------|--------------------------|--------------------------|-------------------------|-------------------------|-------------------------|-------------------------|----------------------------|----------------------------|-------------------------|-------------------------|-------------------------|-------------------------|-------------------------|-------------------------|
| 1A 1 H $1s^1$ | 2A | | | | | | | | | | | 3A | 4A | 5A | 6A | 7A | 8A 2 He $1s^2$ |
| 3 Li $2s^1$ | 4 Be $2s^2$ | | | | | | | | | | | 5 B $2s^2 2p^1$ | 6 C $2s^2 2p^2$ | 7 N $2s^2 2p^3$ | 8 O $2s^2 2p^4$ | 9 F $2s^2 2p^5$ | 10 Ne $2s^2 2p^6$ |
| 11 Na $3s^1$ | 12 Mg $3s^2$ | | | | | | | | | | | 13 Al $3s^2 3p^1$ | 14 Si $3s^2 3p^2$ | 15 P $3s^2 3p^3$ | 16 S $3s^2 3p^4$ | 17 Cl $3s^2 3p^5$ | 18 Ar $3s^2 3p^6$ |
| 19 K $4s^1$ | 20 Ca $4s^2$ | 21 Sc $3d^1 4s^2$ | 22 Ti $3d^2 4s^2$ | 23 V $3d^3 4s^2$ | 24 Cr $3d^5 4s^1$ | 25 Mn $3d^5 4s^2$ | 26 Fe $3d^6 4s^2$ | 27 Co $3d^7 4s^2$ | 28 Ni $3d^8 4s^2$ | 29 Cu $3d^{10} 4s^1$ | 30 Zn $3d^{10} 4s^2$ | 31 Ga $4s^2 4p^1$ | 32 Ge $4s^2 4p^2$ | 33 As $4s^2 4p^3$ | 34 Se $4s^2 4p^4$ | 35 Br $4s^2 4p^5$ | 36 Kr $4s^2 4p^6$ |
| 37 Rb $5s^1$ | 38 Sr $5s^2$ | 39 Y $4d^1 5s^2$ | 40 Zr $4d^2 5s^2$ | 41 Nb $4d^4 5s^1$ | 42 Mo $4d^5 5s^1$ | 43 Tc $4d^5 5s^2$ | 44 Ru $4d^7 5s^1$ | 45 Rh $4d^8 5s^1$ | 46 Pd $4d^{10}$ | 47 Ag $4d^{10} 5s^1$ | 48 Cd $4d^{10} 5s^2$ | 49 In $5s^2 5p^1$ | 50 Sn $5s^2 5p^2$ | 51 Sb $5s^2 5p^3$ | 52 Te $5s^2 5p^4$ | 53 I $5s^2 5p^5$ | 54 Xe $5s^2 5p^6$ |
| 55 Cs $6s^1$ | 56 Ba $6s^2$ | 57 *La $5d^1 6s^2$ | 72 Hf $5d^2 6s^2$ | 73 Ta $5d^3 6s^2$ | 74 W $5d^4 6s^2$ | 75 Re $5d^5 6s^2$ | 76 Os $5d^6 6s^2$ | 77 Ir $5d^7 6s^2$ | 78 Pt $5d^9 6s^1$ | 79 Au $5d^{10} 6s^1$ | 80 Hg $5d^{10} 6s^2$ | 81 Tl $6s^2 6p^1$ | 82 Pb $6s^2 6p^2$ | 83 Bi $6s^2 6p^3$ | 84 Po $6s^2 6p^4$ | 85 At $6s^2 6p^5$ | 86 Rn $6s^2 6p^6$ |
| 87 Fr $7s^1$ | 88 Ra $7s^2$ | 89 †Ac $6d^1 7s^2$ | 104 Rf $6d^2 7s^2$ | 105 Db $6d^3 7s^2$ | 106 Sg $6d^4 7s^2$ | 107 Bh | 108 Hs | 109 Mt | 110 | 111 | 112 | Unknown | 114 | Unknown | ††116 | Unknown | ††118 |

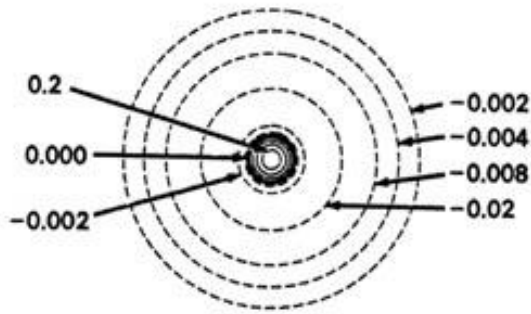
| | | | | | | | | | | | | | |
|---------------------------|------------------------------|-----------------------------|------------------------------|-------------------------|-------------------------|------------------------------|-------------------------|----------------------------|----------------------------|-----------------------------|-----------------------------|-----------------------------|----------------------------------|
| * 58 Ce $4f^2 6s^2$ | 59 Pr $4f^3 6s^2$ | 60 Nd $4f^4 6s^2$ | 61 Pm $4f^5 6s^2$ | 62 Sm $4f^6 6s^2$ | 63 Eu $4f^7 6s^2$ | 64 Gd $4f^7 5d^1 6s^2$ | 65 Tb $4f^9 6s^2$ | 66 Dy $4f^{10} 6s^2$ | 67 Ho $4f^{11} 6s^2$ | 68 Er $4f^{12} 6s^2$ | 69 Tm $4f^{13} 6s^2$ | 70 Yb $4f^{14} 6s^2$ | 71 Lu $4f^{14} 5d^1 6s^2$ |
| † 90 Th $6d^2 7s^2$ | 91 Pa $5f^2 6d^1 7s^2$ | 92 U $5f^3 6d^1 7s^2$ | 93 Np $5f^4 6d^1 7s^2$ | 94 Pu $5f^6 7s^2$ | 95 Am $5f^7 7s^2$ | 96 Cm $5f^7 6d^1 7s^2$ | 97 Bk $5f^9 7s^2$ | 98 Cf $5f^{10} 7s^2$ | 99 Es $5f^{11} 7s^2$ | 100 Fm $5f^{12} 7s^2$ | 101 Md $5f^{13} 7s^2$ | 102 No $5f^{14} 7s^2$ | 103 Lr $5f^{14} 6d^1 7s^2$ |

Atomic Orbitals

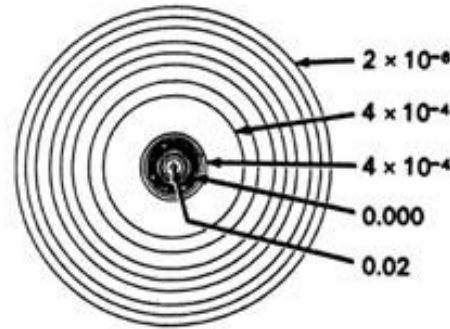


Wavefunction: $\psi_n(r)$

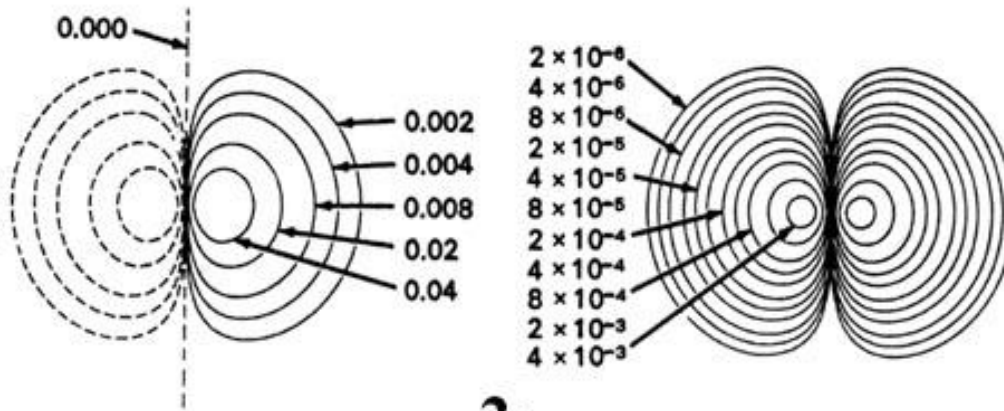
Charge density: $\int \psi_n^*(r)\psi_n(r)dr$



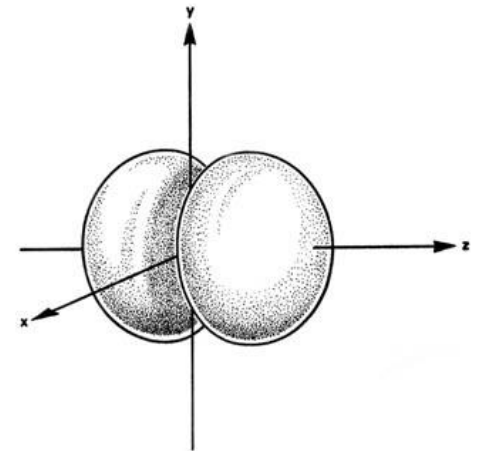
2s



Even parity
 $l=0$



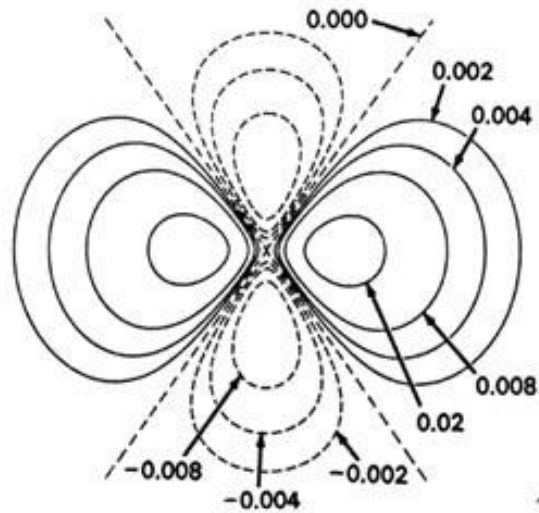
2p



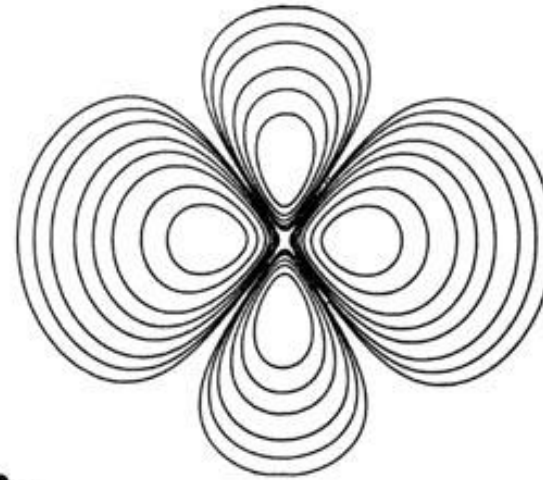
Odd parity
 $l=1$

Wavefunction: $\psi_n(r)$

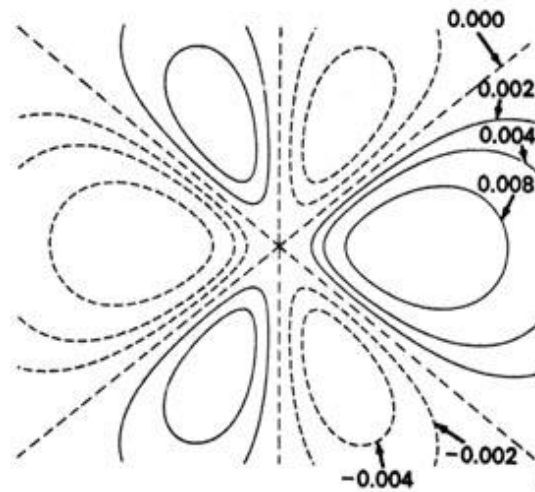
Charge density: $\int \psi_n^*(r)\psi_n(r)dr$



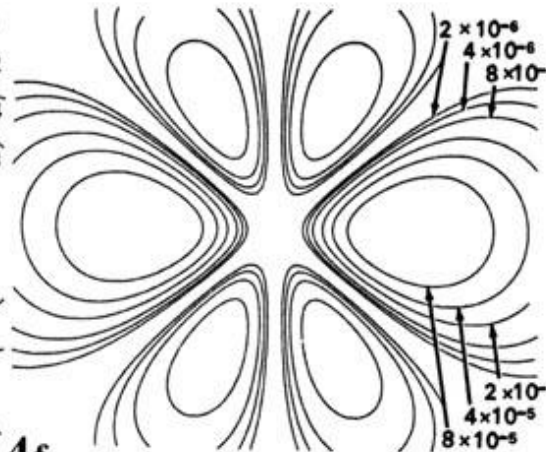
3d



Even parity
 $l=2$

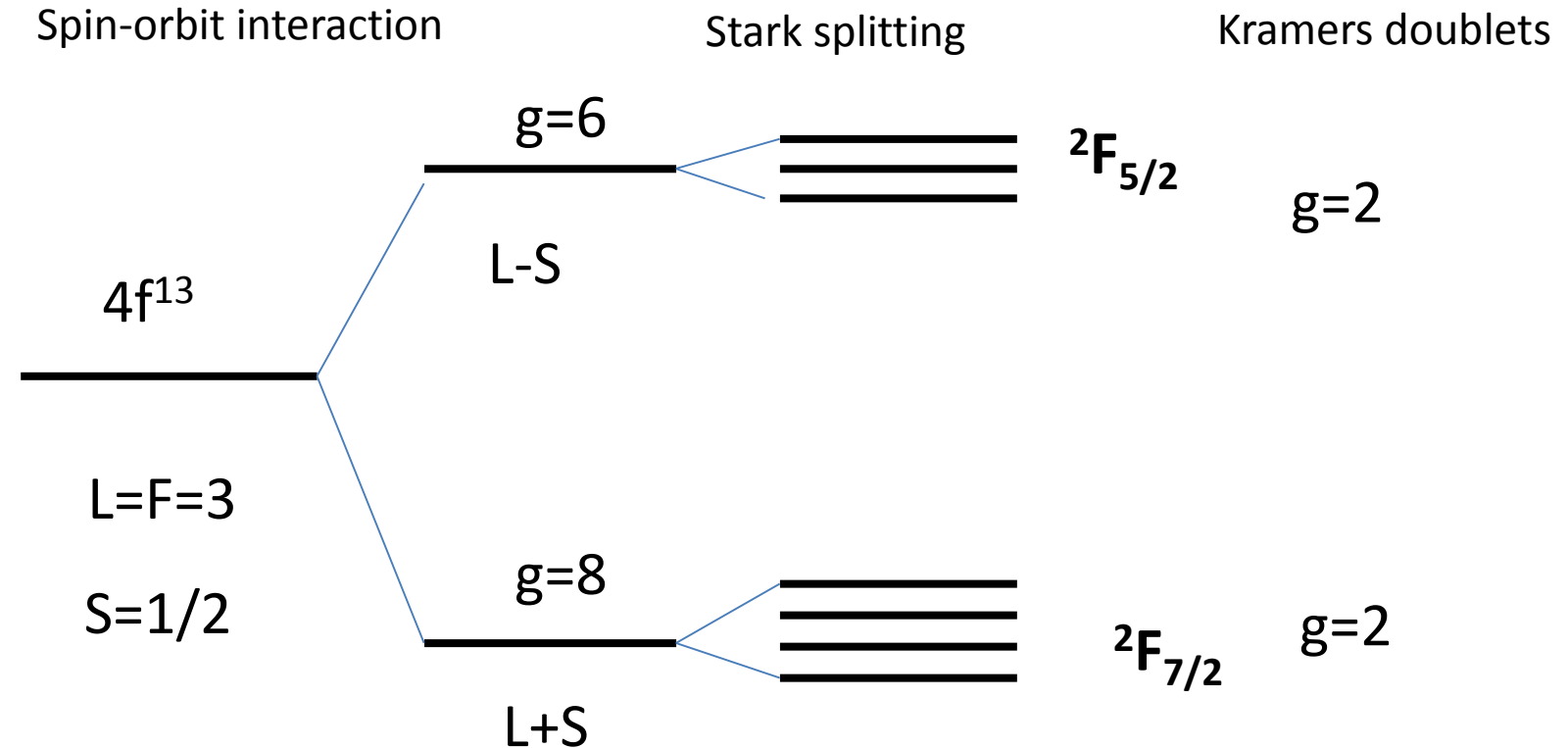


4f



Odd parity
 $l=3$

Yb³⁺ active ion



$$2S+1L_J$$

$$J=L\pm S$$

$$g=2J+1$$

$$L = \sum l; \quad S = \sum s$$

Stark splitting of degenerate states in crystals

Nd³⁺: 4f³

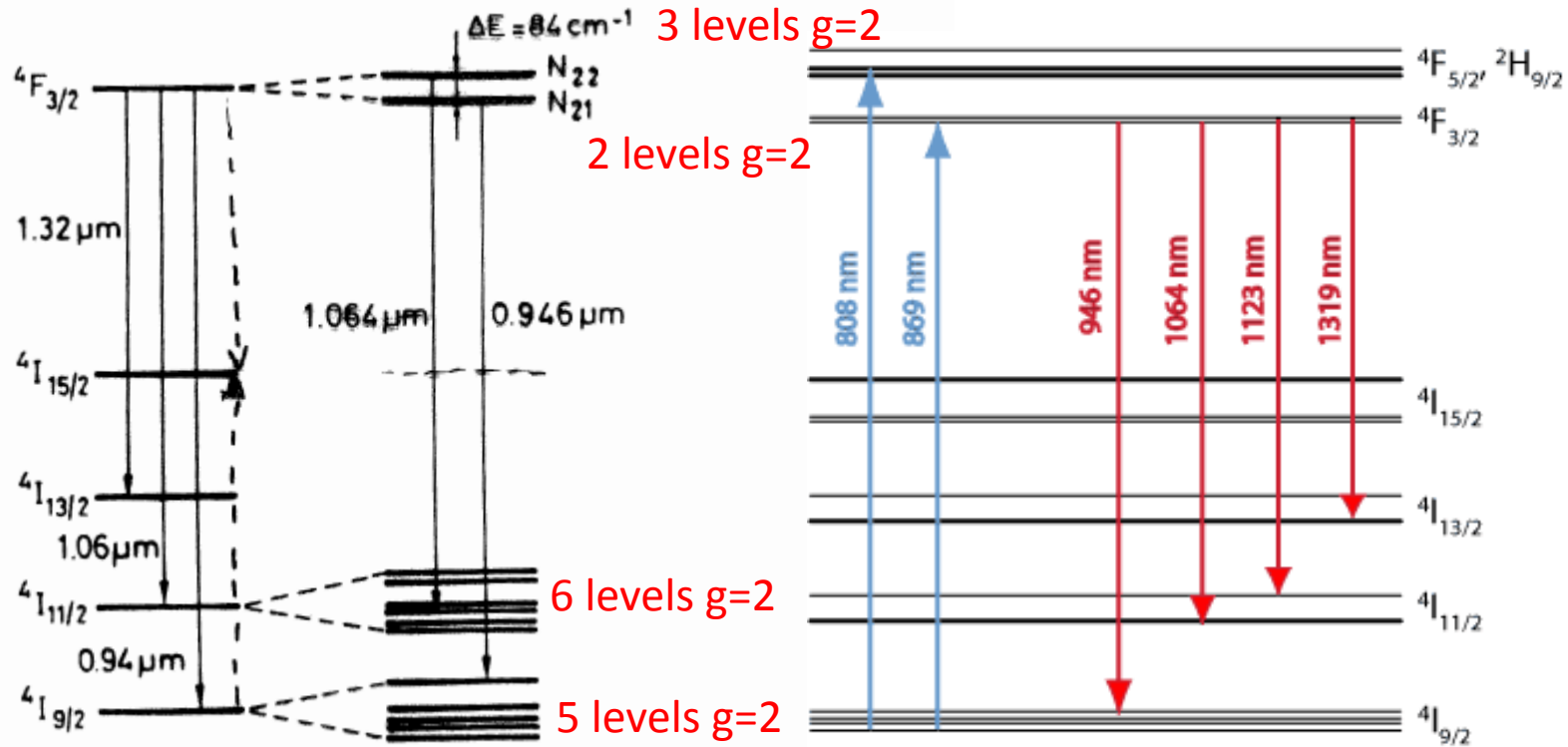
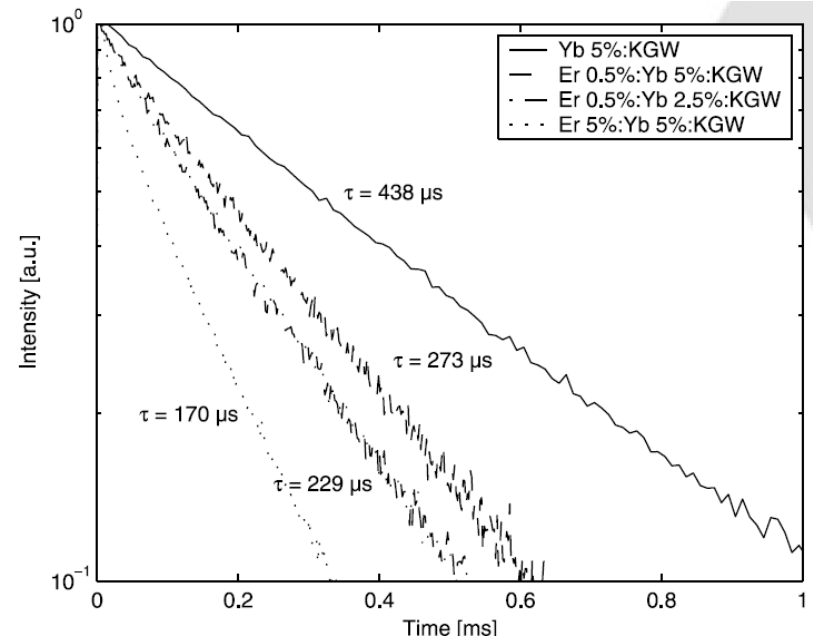
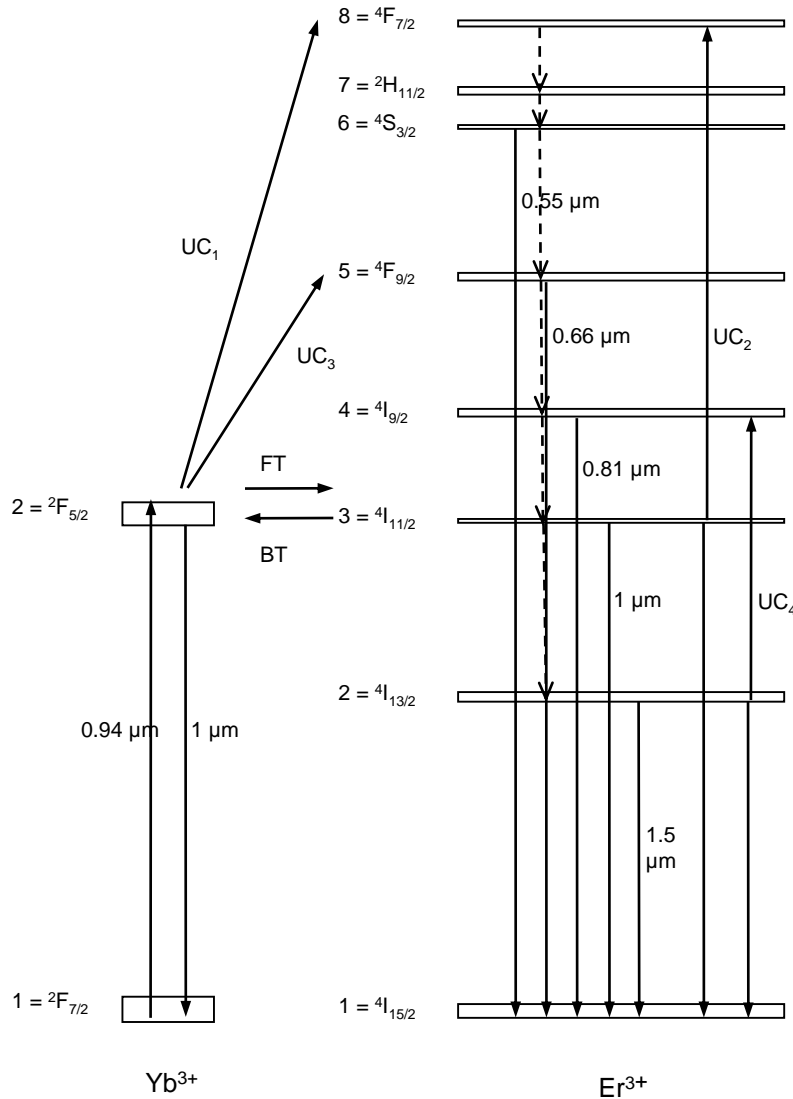


FIG. 2.15. Relevant energy levels for the $\lambda = 1.064\text{-}\mu\text{m}$ laser transition of Nd:YAG laser.

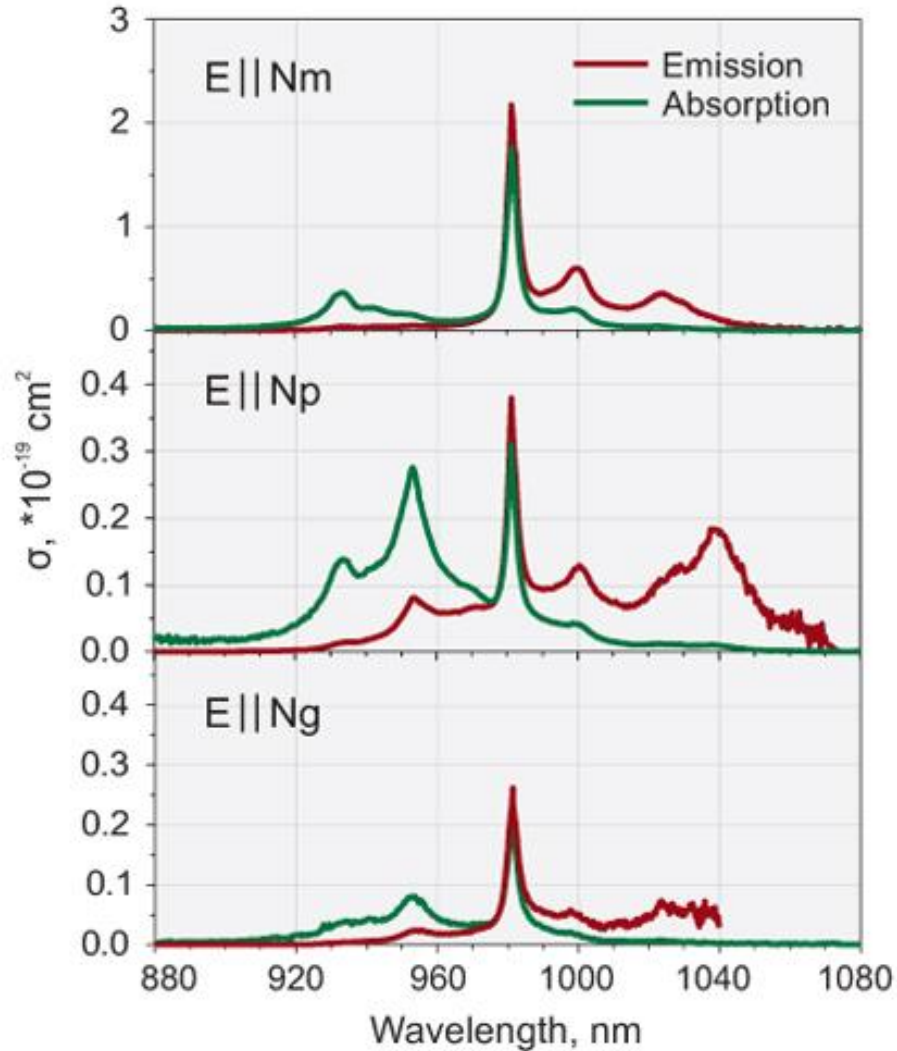
l configurations: $F_{3/2}$: $\downarrow\downarrow\uparrow$ $I_{9/2}$: $\downarrow\downarrow\leftarrow$ $M_{15/2}$: $\downarrow\downarrow\downarrow$

L=3 L=6 L=9

Er:Yb system: resonant excitation transfer



Absorption and emission in closely coupled level systems



Yb:KY(WO₄)₂

Saturation of homogeneously broadened line

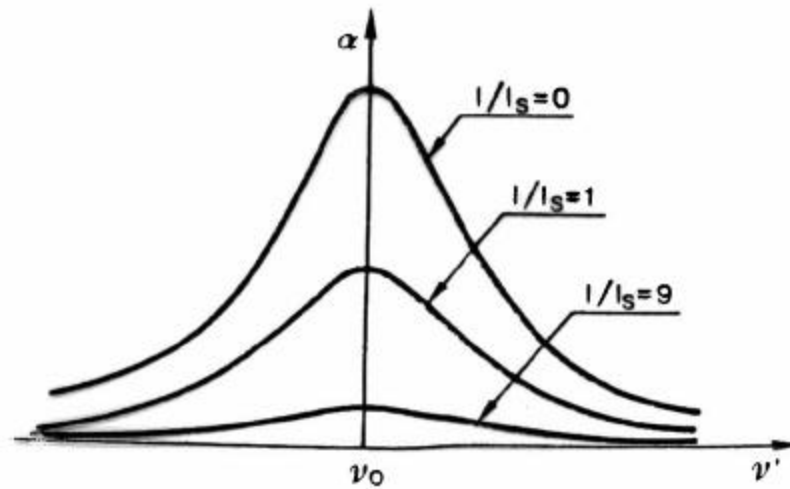
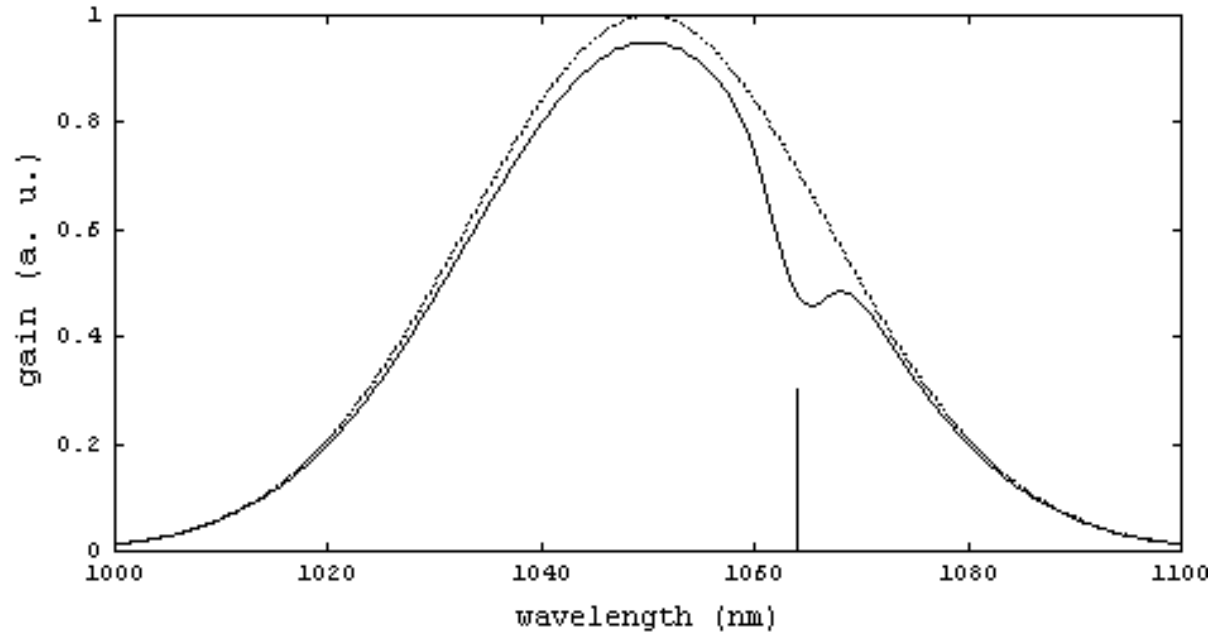


FIG. 2.19. Saturation behavior of the absorption coefficient α versus frequency ν' for increasing values of intensity I of the saturating beam (homogeneous line).

Gain saturation in nonhomogeneous media



Spectral hole-burning

ASE Spectral narrowing

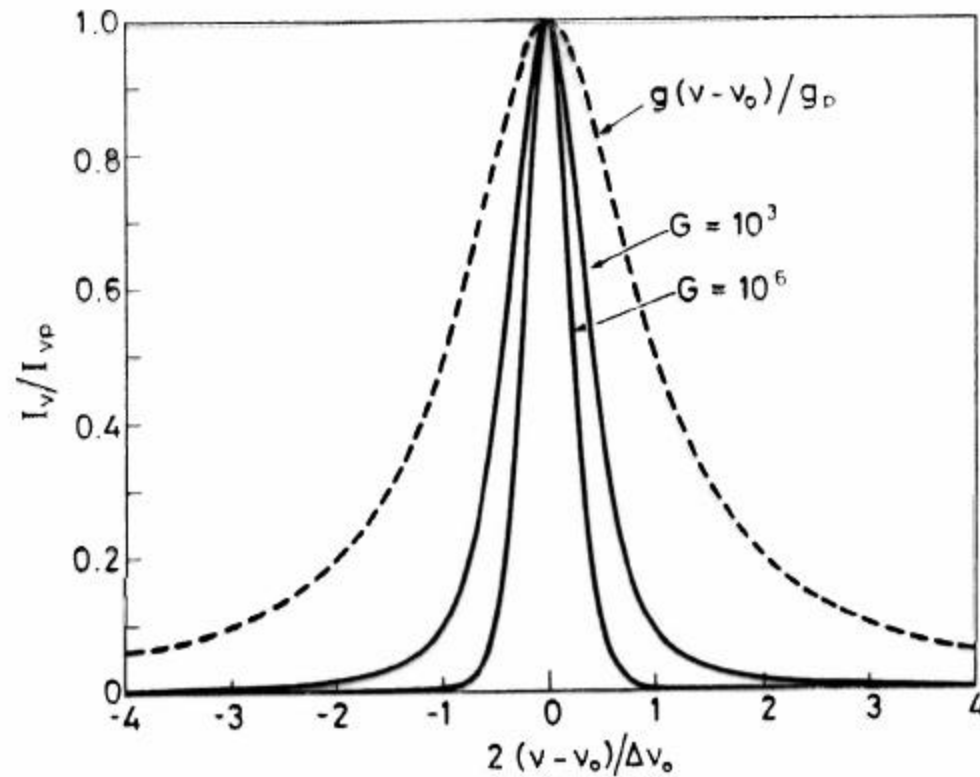


Fig. 2.24. Normalized ASE spectral emission at two different values of the peak, unsaturated, single-pass gain.

Main Keywords

Collisional and multiphonon relaxation

Elastic and inelastic collisions

Cross-relaxation, upconversion, excitation migration and trapping

Degeneracy of electronic levels

Thermalisation of strongly coupled levels

Saturation intensity, saturation fluence

Spectral hole-burning

Problems

2.5, 2.6, 2.8, 2.13, 2.14, 3.2, 3.3,3.4

Examples: 2.10, 3.1